**REVIEW FOR QUIZ: VSEPR, Intermolecular Forces, & Polarity**

1. For the following molecules
2. Draw the Lewis Dot Diagram
3. Name the shape
4. Determine the bond angles
5. Determine the strongest intermolecular force present
6. Classify as polar or non-polar, and show the dipole arrows.

i) SF2

ii) NSH (Nitrogen is the central atom)

iii) SiS2

iv) CHF3

v) CCl4

vi) PCl3

vii) COF2

viii) PCl5